

As-H

$\text{AsH}^+(2)(\Delta fH)$

~~1968~~

594-III-ЛКВ.

Емельянов А.М.

Потенциалы окисления молекул AsH_3 , радикалов AsH_2 и AsH .
- 5с.

AsO₄³⁻ (aq)

Отпуск 35617 | 1991

Касенов Б.К.,

Известие АН КазССР
Серия химическая. 1991,

(АНФ)

№ 5, 82-83

As₂H₄(2) (AfH)

596-III-ТКВ

Меглед В. А.

Пометили и показаны
и этикетки образцов
As₂H₄(2в3), 2с.

$AsH_2^+(2)(\Delta fH)$

~~1968.~~

594 $\equiv$ -III-ПХВ

Емельянов А.М.

Потенциалы окисления молеку-
лы AsH_3 , радикалов AsH_2 и AsH .
-5с

$AsH^-(2) (AfH)$

1968

592 $\approx$ - III - ПКВ

Карачевцев Т. В.

Способы к электроны радикалов
 AsH_2 и AsH , $A(AsH_2)$ и $(\underline{AsH})$. - 4 с.

$\text{AsH}_2^- (2) (\Delta fH)$

~~1968~~
592 Б-III-ПКВ

Карачуев Т.Б.

Сродство к электрону радикалов
 AsH_2^0 и AsH^- , $A(\text{AsH}_2)$ и $A(\text{AsH})$, - 4с.

3484-III

1963

Si_2H_6 , Ge_2H_6 , Ge_3H_8 , Sn_2H_6 , P_2H_4 ,

As_2H_4 , Sb_2H_4 (AlH_3^0)

Saalfeld F.E., Svec H.J.

Inorg. Chem., 1963, 2, 50-3

The mass spectra of volatile ...

u 10

CA., 1963, 58, N 5,

4131d

NOTES ON PAPER

Hf (H_2 st Li H_3 , Li_2 st Li H_3 , 1964
st H_2 , Li H_2) XIII 906

I (st H_2); II (H_2 st Li H_3)

Saunders F. B., McDowell M. V.,
Inorgan. Chem., 1964, 6,
11, 96-98.

let's open

u v

px 1968

BP-8326-XIV

~~РтЗ, AsH₃, H₂S (p, атв, Тв)~~

1967

Зарит А. Д., Красови С. П. XII-233a

Давление пара арсина, фосфина
и сероводорода.

Изв. высших учеб. заведения. Ленин
и хим. технологии. 1967, 10, № 10,
1097-1100.

Р Ж хим., 1968
165845, СХ 1968

5 (ор)

1968

Uzrugze Si, Ge, Sn, P, As, Sb (cb-ba)

XIII 313

Folly W. L., Norman A. D.

Prep. Inorg. React., 1968, 4, 1-58

M, B

CS, 1968, 69, E C 7 0 0 2 7 72737X

AsH

10m. 197321

1984

Kawaguchi K., Hirota E.

лк сакерп,
спусам
носмолк.

J. Mol. Spectrosc., 1984,
106, N2, 423-429.

AsH

1985

3) 11 Б1182. Инфракрасная полупроводниковая лазерная спектроскопия фундаментальной полосы $\text{AsH}(X^3\Sigma^-)$. Infrared diode laser spectroscopy of the fundamental band of $\text{AsH}(X^3\Sigma^-)$. Апасона J. R., Davies P. B., Johnson. S. A. «Mol. Phys.», 1985, 56, № 4, 989—993 (англ.)

$X^3\Sigma^-$

(Vi, M. N.)

С помощью полупроводникового лазерного спектрометра измерено положение 23 линий вращат. структуры фундаментальной полосы в ИК-спектре $\text{AsH}(X^3\Sigma^-)$. Молекулы AsH образовывались в заполненной водородом разрядной трубке, содержащей порошкообразный As в условиях стабилизированного высоковольтного разряда постоянного тока. Для состояния $v=1$ $\text{AsH}(X^3\Sigma^-)$ получены след. значения ν_0 , B , D , λ , γ (в см^{-1}): 2076,874; 6,98918; $3,247 \cdot 10^{-4}$, 58,681; $-0,2573$. В результате совместной обработки полученных результатов и данных по спектру лазерного магн. резонанса $\text{AsH}(X, v=0)$ рассчитаны равновесные спектроскопич. постоянные AsH : $\omega_e=2175$, $\omega_e x_e=49$, $B_e=7,3067$, $\alpha_e=0,2117$, $D_e=8,30 \cdot 10^{-4}$, $\beta_e=3 \cdot 10^{-6}$ см^{-1} ; $R_e=1,52315$ А.

В. М. Ковба

X. 1986, 19, N 11

AsH_n (om 31641) 1988
($n=1-3$) Berkowitz J.,

J. Chem. Phys., 1988, 89,

N 12, 7065-7076.

ΔH_f^0 ,

Photoionization mass
spectrometric studies of
 AsH_n ($n=1-3$).

AsH₂ [om. 31641] 1988
Berkowitz J.,

No, Δ_fH; J. Chem. Phys., 1988, 89,
N12, 7065-7076.

Photoionization mass
spectrometric studies of
AsH_n (n=1-3):

ASH

(DM. 31641)

1988

Berkowitz J.,

J. Chem. Phys., 1988, 89,

Do, ASH;

N12, 7065-7076.

Photoionization mass
spectrometric studies of
 AsH_n ($n=1-3$).

AsH_n^+
($n=1-4$)

1990

Binning R.C., Jr.,
Curtiss L.A.

et al.

J. Chem. Phys. 1990.
92, N6.C. 3688-3692.

( GeH_n^+ ; III)

AsH

AsH₂

memorandum -
cl - la

(42) 12

1993

118: 242028n Equilibrium gas-phase composition and thermodynamic properties including subhydrides in the pyrolysis of arsine and phosphine. Jordan, A. S.; Robertson, A. (AT and T Bell Lab., Murray Hill, New Jersey 07974, USA). *J. Cryst. Growth* 1993, 128(1-4), 488-93 (Eng). A thermodyn. anal. was performed on the pyrolysis products of AsH₃ and PH₃. For the first time were detd. the concn. of subhydrides, (AsH, AsH₂ and their P analogs) resulting from the decompn. In the case of AsH₃ and the corresponding subhydrides, also were detd. the thermodyn. properties from critically assessed or estd. spectroscopic data by means of statistical thermodyn. A novel free-energy minimization technique permits the calcn. of species concns. that differ by orders of magnitude. In applying the mass conservation constraints the authors differentiated between flow systems in metalorg. chem. vapor deposition (MOCVD) and crackers with effusive flow through an orifice in metalorg. mol. beam epitaxy (MOMBE). The authors have found that significant concns. of H, As, P, and AsH are generated in MOMBE crackers. In low pressure MOCVD of quaternary III-V alloys, PH₂ is the most important intermediate species. The calcd. concns. are in reasonable agreement with mass spectrometric measurements both at low and high pressures.

C.A. 1993, 118, N 24 PH₂, P.H⁻

1993

 A_3H A_3H_2 A_3H_3 $(\Delta f, \Delta S)$ $(+2) \quad \Delta$  PH, PH_3 C.A. 1993, 119, N 12

119: 130402u Equilibrium gas-phase composition of cracked arsine and phosphine. Robertson, A., Jr.; Jordan, A. S. (Eng. Res. Cent., AT and T Bell Lab., Princeton, NJ 08540 USA). *J. Vac. Sci. Technol., B* 1993, 11(3), 1041-4 (Eng). A thermodyn. anal. for the pyrolysis products of AsH_3 and PH_3 was performed. For the 1st time the authors detd. the concn. of subhydrides, (AsH , AsH_2 , PH , PH_2) resulting from the decompn. In the case of AsH_3 and the corresponding subhydrides, the authors also detd. the thermodyn. properties (free-energy function, entropy, etc.) from critically assessed or estd. spectroscopic data from statistical thermodyn. A free-energy minimization technique permitted the calcn. of species concns. that differ by orders of magnitude. The mass conservation constraints formulated here are applicable to crackers with effusive flow through an orifice as used in metallorg. MBE and hydride MBE. The calcd. concns., including the dimer/tetramer ratios, are in reasonable agreement with mass spectrometric measurements.

AsH₂
AsH₂-

1997

Mayer P.M. et al.,

Int. J. Mass Spectrom.
Ion Processes 1997,
(Δ+H) 167/168, 689-696.

(coll. BeH₄;I)