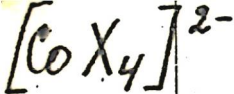


Со - залотениды



X = Cl, Br

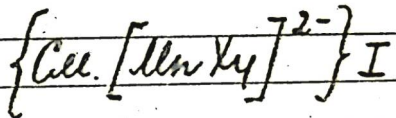
непрямая
связь

Paoletti P.
Sabatini A.

1964

Proc. Int. Conf. Coord. Chem.,
8th, Vienna,

1964, 25-9



1970

Collatz (aq)

57894b Dissociation of cobalt(II) halide solvates. Bkouché-Waksman, Itka (Lab. Rech. Chim. System., Paris, Fr.). C. R. Acad. Sci., Ser. C 1970, 271(10), 581-4 (Fr). The dissocn. pressures of the following solvates are tabulated for the indicated temp. ranges: $\text{CoCl}_2 \cdot 2\text{H}_2\text{O}$, 36.5-119.7; $\text{CoCl}_2 \cdot \text{H}_2\text{O}$, 36.5-158.0; $\text{CoCl}_2 \cdot \text{MeOH}$, 24.6-125.4; $\text{CoBr}_2 \cdot \text{MeOH}$, -1.7 to 55.7°. No other MeOH solvates were found at these temp. ranges. The changes of free energy, enthalpy, and entropy for each equil. are calcd. The enthalpy change is independent of temp.

A. J. Miller

C. A. 1971. 74. 12.

1987

Залогенедн,
идридн
и
дейтеридн
Co, + и-ионив

Jackson C.R.,
Chase M.W.,

Report 1987, NBS/TN-1244,
Order No PB88-20/603,
53pp.

термод. св-ва,
визмограф.

(см. справочники,
I (общ. вопросы))

1979

Co-raznomeniya

Bartovska L. et al

SB. Vys. Sk. Chem. Technol.

Praxe, Phys. Chem. 1979,

N3, 55-144

•
Cil Fe

ozor

T.g.

cb-ba