

MgH



MgM.

Вср-6927-VI /1940
Stevenson D.P
J. Ch. Phys. 8, No стр 898

S^0
 $S_{298,1}$

Заметка об энтропии
газообр. 2^е атомных
гидридов и галогенидов
(Менделеевские расстояния
и частоты)

$$\chi_e = 1.741 \text{ Å}^3$$

$$w_e = 1350(2)$$

см K₁M



Hg H
Hg D
Hg T

coll. vol. 10000 in abstract 1961

Haar L., Friedman A.S.,
Beckett C.W.

Ideal Gas Thermodynamic
Functions and Isotope
Exchange Functions for
the Diatomic Hydrides,
Deuterides, and Tritides.

NBS, Monograph 20, Washington,
1961.

1985

HgH (gas)

JANAF

T. ϕ .

100 - 6000°K

V 5940

1966

MgH, B₂, Na₂, Zn₂, K₂, C₂, Γ₂ и HF, Γ-анод.
(термохимическ. сп-ции)^{Feber}

Feber R.C., Herrick C.C.,

U.S. Atom. Energy Comm. LA-3597,
Avail Rep. in., CFSTI, 1966, 70 pp

10

CA, 1968, 68, №4, 166429

HgH

omnick 3392

1967

Felber R.G.,
Herrick C.C.

m.g.p.

Report written
September 1966, distributed
1967, LA-3597 UC-4 Chem.

TID-4500.

An improved calculation
of the ideal gas thermodyn...

H₂H
(Ideal gas)

100-6000°K
(1963)

7/11/71
Tuyg

1971

HgH (e.)

Barin J; et al.

1973

298-2000

v. I, p 340

(see. AgF; I)

HgH⁺ omuck 3176 1974

Schneider J.

Z. Phys. Chem.

m.g. p.

стат. сум. 1974, 255, 986-996.
номинант

● (стат. сум.
II)