

CS₂



И-9.

Evans W.H., Jacobson R., 1955

ф-444

Munson P.R., Wagman D.D.

Cs₂

Y. Res. NBS. 1955, 55, N2, 83.

9025000

возможности с Na_2

Вар-117

$T^{\circ}R$	Q	S
298.15	-59,019	67,855
500	-63,623	72,596
1000	-69,91	79,09
1500	-73,66	83,01

CS₂

1956

Stull D.R., Sinke G.C.
The Thermodynamic Properties
of the Elements

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v. 8p 1955.

C_2
(Ideal gas)

100-6000°K
(1965)

JANAF
Thg

1971

Cs₂

omitted 1973

1973

T.G. 90

Reiter F.W.,

Z. Naturforsch., 1973,

28a, No 10, 1676-1686.

● (cur. Li₂; II)

CS₂

Bp - 1005 - $\bar{X}$ B

1977

Behrens R. C.

Woodrow H. O. et al

m.g. g

J. Chem. Thermodyn.
1977, 9 (11) 1035-44.

(em. C₃; I)

$\text{CS}_2(g)$

1977

Barin J, et al

m.g.p.

mon II, mp. 215

298-2000

● (see Ag) I

Cs₂

ommuch 6648 4978

Hessel

m. g. g.

to be published

G₂(2)

1982

Pan Kratz L. B.

Thermodynamic Properties
(298-2000) of Elements and Oxides
USA Bur. Mines Bull.
672.

● (y Megbegeho)

G₂ (P) (om-24103) 1985

meperog.
go-uu,
Do

Gurvich L.V., Yozish V.S.,
Khandamirova N.E.,
Yungman V.S.,
Handb. Thermodyn. and
Transp. Prop. Alkali Metals,
Oxford, 1985, 577-627.

Ga⁺(17)

(m. 24103)

1985

Gurvich L.V., Yorish V. S.,
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перевод.
90-44,
Do

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С2 (9)

1985

YANAF

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