

FeBr



FeBr<sup>2</sup>

OM 34712 ]

1939

Mémoire coll. P.,

Ann. Phys. (Fr), 1939, 12,  
5-87.

crekmp

Recherches sur les décharges de  
haute fréquence et leur appli-  
cation à la spectroscopie

moleculaire.

Feb 2

1959

11-1308  
New band systems of FeBr in the visible region. S. Paddi Reddy (Andhra Univ., Waltair). *J. Sci. Ind. Research (India)* 18B, 188-90(1959).—Four band systems attributed to the LiBr mol. were observed in the region 5720-6410 Å. Pure, anhyd. FeBr<sub>3</sub> was excited with a high-frequency discharge from a 100-w. oscillator and a heavy current discharge from a 2000-v. d.c. generator. Starting from the shorter wave length side, the bands are designated as systems I, II, III, and IV. In each of the 1st 3 band systems, only 2 sequences were identified, while in the 4th, 3 sequences were observed. The vibrational analyses indicate that systems I, II, and IV have a common lower state, which is probably the ground state of the FeBr mol. Systems I and II, which consist of single-headed bands, may belong to a transition of the type  $\Sigma - \Sigma$  with multiplicity 2 or 4. The bands of the systems III and IV show P and Q heads. In the 0,0 band of system IV, the 0 head is identified in addn. to the P and Q heads. Each of the systems III and IV may belong to a transition of the type  $\pi - \Sigma$  with multiplicity 2 or 4.

T. P. Gross

C.A. 1961, 55 19.  
18292 ab.

FeBr

~ 1968

Amherst 160P

J.W.Hastie J.L.Margrave.

"Dep.of Chem., Rice

University Houston, Texas 77001.

p I-50.

1970

FeBr

10 Д336. Новая система полос молекулы FeBr.  
Krishna Rao N. V., Tiruvenganna Rao P.  
 A new band system of the FeBr molecule. «J. Phys.»  
 (Gr. Brit.), 1970, B 3, № 5, 725—728 (англ.)

новая  
система  
полос

Спектр излучения молекулы FeBr возбуждался с помощью ВЧ-разряда в кварцевой кювете, содержащей нагретый железный порошок и пары Br. На спектрограммах, полученных с дисперсией 1,25 Å/мм, в области 3650—3760 Å обнаружена новая система полос с фиолетовым оттенением, обусловленная переходом  $a^4\Pi - ^4\Sigma$  и аналогичная II системе FeCl в области

09. 1970

108

3500—3650 Å. Положение наиболее резких кантов, образованных  $P_4$ -ветвями, может быть представлено в виде:  $v_v' v'' = 27056,8 + 315,1(v' + 1/2) - 0,4(v' + 1/2)^2 - 304,3(v'' + 1/2) + 0,85(v'' + 1/2)^2$ . Приведена таблица наблюдаемых кантов полос; правильность классификации подтверждается изотопич. сдвигом полос.

В. Александров

FeBr

1970

(50458h) Newband system of the iron bromide [FeBr] molecule.  
Rao, N. V. Krishna; Rao, P. Tiruvenganna (Dep. Phys., Andhra  
Univ., Waltair, India). *J. Phys. B* 1970, 3(5), 725-8 (Eng).  
The emission spectrum of the FeBr mol. is excited in a high-fre-  
quency discharge tube through a mixt. of heated Fe metal and  
Br vapor. The spectrum which has been photographed under  
high dispersion (1.25 Å/mm) reveals a new system of violet-  
degraded bands in the region  $\lambda\lambda$  3650-3760 Å. The system is  
interpreted as arising from a  $^4\Pi$ - $^4\Sigma$  transition analogous to the  
system  $II$  of FeCl in the region  $\lambda\lambda$  3500-3650 Å, analyzed pre-  
viously. The  $P_4$  heads can be represented by the following quan-  
tum formula  $v, v', v'' = 27056.8 + 315.1(v' + 1/2) - 0.40(v' +$   
 $1/2)^2 - 304.3(v'' + 1/2) + 0.85(v'' + 1/2)^2$ . The vibrational  
assignments of the band heads are confirmed from a study of the  
obsd. Br isotope effect.

RCBS

C.P. 1970. X3.10

Fe	Br	Химическая перегруппировка в реакции $\text{Fe}^+$ с $\text{SF}_6$
		1974
название		Новое соединение танталового кислорода.
авторы		Brinton Robert K.
		"J Chem Soc Faraday Trans"
		"1974, Part 2, 70, N2, 203-214 (авт.)
x. 1974. N 17		[an Fe Cl; III]
		3ак. 247

Fe-Bz

OTT. 4824

1985

Kerr J. A., et al.

(Q)

Handbook Chem. Phys.,  
55 th Ed., 1974-75.

FeBr<sup>+</sup>

1994

Hrušák Jar, Fiedler A,  
et al.

8th. Int. Congr. Quar-

m. n. turn Chem., Prague,  
Jeře 19-23, 1994. Book  
Abstrs. [Prague], [1994].

- C. 103.

(Cet.  $\text{FeF}^+$ ; III)



Feß  
u. Mitarbeiter, Ch. W., Y.;  
Höfer 1996

Chem. Phys., 1996, 211(1,2,3)  
Do, 163-169  
meop.

Bauer

(all. FeF<sub>1</sub>III)