

Ag-S

V 2625

1960

Ag₃SBr (re)

Ag₃SJ (re)

Reuter B., Hardel K.

Angew.Chem., 1960, 72, N 4, 138-9

Silbersulfidbromid und silbersulfidjodid.

PJX, 1960, 76278

ML.

1972

Ag 25

ammuer 3133

Smoes S, et. al.

Bull. Soc. Crim Belg.

1972, 81, 45-56

(do)

Ag 25

summer 6287 1978

Romand M., et al.

promos.

Once-cueup J. Electron Spectrosc.
and Relat. Phenom.,
1978, 13, 229-42.

Ag-Scoregan.

1981

Tossell J.A., et al.

Heppner Inorg. Chem. 1981, 20,
Cv834.

N 10, 3333 - 3340.

(, Cu-Cloeg; (11))

AgS

[Om. 16806]

1983

Koniam.
noem.

99: 61008z Prediction of vibrational constants for gas-phase heteronuclear diatomics and atoms adsorbed on surfaces. Goodfriend, P. L. (Dep. Chem., Univ. Maine, Orono, ME 04469 USA). *Chem. Phys. Lett.* 1983, 98(1), 37-9 (Eng). An empirical relationship is presented for predicting vibrational consts. for gas-phase heteronuclear diat. mols. and atoms adsorbed on surfaces. Excellent agreement with exptl. values is obtained. The method is used for predicting the unknown vibrational consts. of AgS and CTe.

(H) ☒

C.A. 1983, 99, 18

Ag 3

[OM. 35679]

1990

Kymna. Baeschlicher Ch. W., Jr.,
gyneth. Patridge H., et al.,
Eccm.
ad initio Chem. Phys. 1990, 148,
pauet 57-68
N11,

Ag₂S

1992

Dixon D.A., Cole J.L.,

Chem. Phys. Lett., 1992,
189 (4-5), 390-4.

осн. сост.,
структура,
D_i

(see Ag₂O ● ; III)

(Agg. 3) n

[Cm. 41222]

1998

N=1/2 A.A. Bagatur'yants, ~~et~~

A.A. Sazonov et al.

ad intro
part

J. Chem. Phys., 1998,
109, N 8, 3096-3107

AgS

2001

Legge, F. Sue et al.,

протерка
метода
расчета
DFT

J. Phys. Chem. A 2001,
105 (33), 7905-16

в различных
приемах.

● (Calc. Chem, III) ?

Ag₂S

(DM 41464)

2002

Ag₂Se

Yongqiang Zhao et al.,

Ag₂Te

J. Mat. Struct. (Theochem),

ab initio
prism

2002, 587, 43-48.