

Ng-S

$MgSiO_3$

БФ-1361-IX

1963

Leconte J. et al.

Comp. Rend. 1963

(J; ступня)

257,326-5

Alg S<sub>2</sub>

1976

Чаркин О. П. и др.

М. иерр. хилл. 1976,  
21, №8, 2022-9,

(переч.  
от. сир)

(all set<sub>2</sub>) III

1976

alg S<sub>2</sub>  
alg O<sub>1</sub>S

Решов М. А. и др.

рук. ген. б ВУХУИ  
3685 - 70 ДЕП

(расчет)  
эн. суп.)

(см. Вет) III

$MgS_2$

Ryabov M. A, et. al. 1977

кв. чер.  
расчет  
симметрии  
структуры

Zh. Fiz. Khim 1977,

51(3) 770 (Russ)

(см BeF, III)

Mg SH  
Mg SH<sup>+</sup>  
Mg(SH)<sub>2</sub>

кб. мех.  
расчет

номер 7042 1978

Pappas, Jan A.

J. Am. Chem. Soc. 1978,  
100(19), 6023-7

сч. Li SH - III

Mg<sub>2</sub>S

Оммух 14384,

1982

Martin F.P., Schaber  
H.,

спектр  
в  
матрице

Spectrochim. acta,  
1982, A38, N 6, 655-  
-660.

Alg S<sub>2</sub>

[Ummuck 14334]

1982

Martin T. R., Schaber  
H.,

спектр  
в  
матрице

Spectrochim. acta,  
1982, A 38, N 6, 655-  
- 660.

S Mg S

(от. 27553)

1987

Зюбина Т.С., Царкин О.П.,  
Зюбин А.С.,

струк-  
туро  
и  
стабильн.

Ис. Меорган. химии,  
1987, 32, n 11, 2616-  
● - 2624.

SMgH<sub>2</sub> Simandiras E. A., 1991  
Nicolaidis C. A.

u. n. Chem. Phys. Lett. 1991.  
185. N 5-6. C. 529-534.

(Cui. ● MgH<sub>2</sub><sup>2+</sup>; III)

$Mg_2S_2$

$Mg_3S_3$

1992

Al-Laham Moham-  
mad A., Trucks Gary W.  
et al.

красоты,  
их структур.  
и устойчив.

J. Chem. Phys. 1992,  
96, N 2. C. 1137-1149.

(ссылка:  $Al_nP_n$ ; III)

HS Mg H<sub>2</sub>  
FS Mg H<sub>2</sub>

Зюбина Т. С.  
Чаркин О. Я.

1992

ab initio  
расчет

Ж. Неорганич. химии.  
1992. 37, № 9. 865-  
-871.

(сер. ● MOBH<sub>2</sub>; III)

Mg<sub>4</sub>Si<sub>4</sub>

1993

Al-Zakam Mohammed  
A., Raghavachari K.

еетрехт.,  
еиабилвн.,  
теор,  
раерет

J. Chem. Phys. 1993,  
98 (11), 8770-6.

●  
(see. Ga<sub>4</sub>As<sub>4</sub>; III)

$Mg_3S^{2-}$

1993

Boldyrev A.I.,  
Simons J.

u.n..

J. Chem. Phys. 1993, 98,  
N6. C. 4745-4752.

● (ex:  $C_2^{2-}$ ; III)

$SMgS^{2-}$

1993

Bolchuzer A.I.,  
Simons J.

u.n. J. Chem. Phys. 1993, 98,  
NB.C. 4745-4752.

● (cell.  $C_2^{2-}$   $\frac{1111}{1}$ )

MgS<sup>2-</sup>

1993

Boldyrev A.I.,  
Simons J.

u.n. J. Chem. Phys. 1993.  
98, N 6. C. 4745-4752.

● (cu. C<sub>2</sub><sup>2-</sup>; n)

Mg<sub>4</sub>Si.

(DM. 38410)

1993

Mohammad A. Al-Zakam  
and Krishnan Raghavachari,

(U.N.) J. Chem. Phys., 1993, 98  
(11), 8770 ● - 8776

Alg 34

[Om. 40693]

2001

mm-роковой  
спектр

Amine Taleb-ben-  
dial, Darren Cho-  
nick,

Chem. Phys. Lett.  
2001, 334, 195-199

F: MgSH

P: 3

134:258585 Millimeter-wave spectrum of MgSH. Taleb-Bendiab, A.; Chomiak, D. Stencie-Institute for Molecular Sciences, National Research Council of Canada, Ottawa, ON, Can. Chem. Phys. Lett. (2001), 334(1,2,3), 195-199. in English.

The pure rotational spectrum of MgSH was obsd. for the 1st time, using millimeter-wave absorption spectroscopy in the 280-365 GHz region. This short-lived gas phase free radical was produced in a high-temp. cell by the reaction of Mg metal vapor with H<sub>2</sub>S in the presence of an elec. discharge. The spectrum was analyzed using an S-reduced asym. rotor Hamiltonian to obtain mol. parameters, including rotational, centrifugal distortion, and spin-rotation consts. The anal. of the spectrum indicated that MgSH is bent, in agreement with an ab initio calen. The Mg-S-H angle obtained from the anal. was  $87.5 \pm 6.7^\circ$ , as compared to a calcd. ab initio value of  $91.1^\circ$ .