

R.R - 7

1973

ReO₃I

141587-u Rhenium iodide oxide [ReO₃I]. Formation, mass spectrum, ionization energy, and enthalpy of formation. Rabeneck, Helmut; Rinke, Klaus; Schaefer, Harald (Anorg.-Chem. Inst., Univ. Muenster, Muenster, Ger.). *Z. Anorg. Allg. Chem.* 1973, 397(2), 112-16 (Ger). In the evapn. of a mixt. contg. ReO₃ and AgI, ReO₃I was obsd. in the mass spec-

ΔH_f trum. The ionization energy of the mol. is 10.9 ev, and the enthalpy of formation $\Delta H^\circ_{298} = -106.1$ kcal/mole.

8/8
- 25

C.A. 1973, 78 n22

(+) ◉

☒

1973

 Re_3I_9 $\text{Re}_3\text{Cl}_{9-n}\text{X}_n$ $X = \text{Br}, \text{J}.$

113905u The gaseous rhenium chloride, bromide, and iodide molecules Re_3Cl_9 , Re_3Br_9 , and Re_3I_9 . Halogen exchange and fragmentation. Schaefer, Harald; Rinke, Klaus; Rahmbeck, Helmut (Anorg.-Chem. Inst., Univ. Muenster, Muenster, Ger.); *Z. Anorg. Allg. Chem.*, 1973, 403(1), 23-34 (Ger.). The mass spectrum of Re_3I_9 was reported and compared with those of Re_3Cl_9 and Re_3Br_9 . The fragmentation corresponds with the stability of the bonds Re-Cl, Re-I, and Re-Re. In the mixed halides $\text{Re}_3\text{Cl}_{9-n}\text{X}_n$ ($X = \text{Br}, \text{I}$) the obserd. abundance of mols. with $n = 0-9$ may be described statistically using the same wt. for all 9 positions of halogen. No difference of the 3 bridging halogens and the 6 terminal halogens can be seen regarding the halogen exchange at $3(\text{R})^\circ$.

massenpekt

C.A.1974.80.N20

40228.3809

Ch, TE

 Re_3J_9

48536

1974

У-3816

Schäfer Harald, Rinke Klaus, Rabeneck Helmut. Die Gasmoleküle Re_3Cl_9 , Re_3Br_9 und Re_3J_9 - Halogen austausch und Fragmentierung. "Z. anorg. und allg. Chem.", 1974, 403, № 1, 23-34 (ном., рез. англ.) 0054 инк

041 042

046

ВИНИТИ

20103.4234

Ex-Ch/XKB--z,
Ch, TC

Re - γ ^{29932.} (11)

1976

X9-16105

Fergusson J. E., Revoldt P. F.

Tertiary phosphine complexes of rhenium
a spectroscopic study. "J. Inorg. and Nucl.
Chem.", 1976, 38, N 12, 2231-2237

(англ.)

07.83 чик

759 764

ВИНИТИ

ReO_3I (ray) Rabeneck Helmut; 1973
et al.

(J)

"Z. Anorg. und allg. Chem."
1973, 397, N2, 182-116.

● (cu. ReO_3I ; I)