

CH₂

CHI₂

~~CHI₃~~

LiI

Cl₃

(Li)

(u.k. chem)

1972

132702e Matrix infrared spectra and bonding in the di- and triiodomethyl radicals. Smith, David W.; Andrews, Lester (Chem. Dep., Univ. Virginia, Charlottesville, Va.). *J. Phys. Chem.* 1972, 76(19), 2718-26 (Eng). The codeposition of either CHI₃ or Cl₄ at high diln. in Ar with ⁶Li or ⁷Li produced new ir absorptions which are attributed to LiI and the CHI₂ or Cl₂ free radicals, resp. The identity of the CHI₂ radical is confirmed by the reactions of CDI₃ with Li and CHI₃ with Na atoms. Assignments are made to the antisymmetric HCl valence angle bend and to the antisymmetric C-I stretch of CHI₂. The pyrolysis of Cl₄ also produced Cl₂; an assignment is made to the antisymmetric C-I stretch. The high values of F_r-F_{rr} in CHI₂ (2.54 ± 0.03 mdyne/Å) and Cl₂ (2.13 mdyne/Å) indicate considerable π -bonding in the CHI₂ and Cl₂ free radicals.

C.A. 1972, 77, 20

CH₂

Smith D.V.,

1972

и К
матрица

Andrews L.

J. Phys. Chem, 1972, 76, 2718

CH₂ молек. V₃-симм. CH₂ деф. } не имеет за
C_{2v} V₆ неплоск. деф. } идеальной
рабочей длины

V₄ асимм. CH₂ деф. 1106,4 см⁻¹.

V₅ C-У вал. колеб- асимм. 716,4 см⁻¹.
stretch. наиболее вращ. в CH₂.

$$F_{44} = \cancel{4 \times 10^8 \text{ dynes/cm}^2} \cdot 2,537 \cdot 10^5 \text{ g/cm}^2$$

$$F_{45} = 0,243 \text{ millidynes/radian}$$

$$F_{55} = 0,395 \text{ millidynes} \cdot \text{\AA} / \text{radian}^2$$

$$(\text{CH}_3 \quad F_{55} = 0,413 \text{ mdy} \cdot \text{\AA} / \text{rad}^2)$$

СМІ₂

1989

Суцаревскі Ч. В.,
Резерко С. М. и др.

Оптика и спектро-
метрия. 1989, 66, N2.
С. 475-477.

и.п.

(см. [СМІ]; III)