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Con

1=3,4,5

(meop. pacien)

105: 197413c Analysis of the reactivity of small cobalt clusters. Rosen, A.: Rantala, T. T. (Dep. Phys., Chalmers Univ. Technol, Swed.). Z. Phys. D: At., Mol. Clusters 1986, 3(2-3), 205-9. (Eagl.) The electronic structure was calcd. of small Co clusters within the local spin d. approxn. by using the LCAO method. The calcus, were made for simple geometries with the optimized no. of interat. bonds both for the bond length of Co_2 and the bulk metal. The Fermi energy is smaller for Co_3 clusters with n=3, 4, 5 and n > 10 that for other clusters. The variation of the Fermi energy with the cluster size correlates with the obsd. H_2 tendency for chemisorption, a found for Co_3 clusters in a supersonic beam. Magnetic moments are somewhat smaller for these active clusters. The lowest unoccupied levels of majority spin appear close to the highest occupied levels of minority spin which is not the case for the inert clusters.

C.A.1986, 105, N22

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measured with photedissociation. Zheng, Lansun; Yang, Shihe (Postdoctoral Stn. Phys. Chem., Xiamen Univ., Xiamen, Peop. Rep. China). Wali Huaxue Xuebao 1990, 6(3), 272-6 (Ch). The binding energies of transition metal (Co, Nb, Fe, Ni, V, Ta, Pd) clusters have been measured as functions of cluster size and compn. with cluster size up to 20 atoms. The photodissocn. measurement showed that the binding energy is roughly const. with different cluster sizes but differs with different cluster compns. For almost all clusters measured, the primary one-photon fragmentation pathway is always found to be loss of a single metal from the cluster.

113: 65654n Binding energies of transition metal clusters

C.A.1990, 113, N8

1=2+18

120: 174563x Collision-induced dissociation of Con+ (n = 2-18) with Xe: bond energies of cationic and neutral cobalt clusters. dissociation pathways, and structures. Hales, David A.; Su, C. X .: Li, Lian; Armentrout, P. B. (Dep. Chem., Univ. Utah, Salt Lake City, UT 84112 USA). J. Chem. Phys. 1994, 100(2), 1049-57 (Eng). The kinetic energy dependence of collision-induced dissocn. (CID) of Con+ (n = 2-18) with xenon is studied by using a guided ion beam mass spectrometer. Examn. of the general dissocn. behavior over a broad collision energy range shows that cobalt cluster cations dissoc. exclusively by loss of single atoms (cluster evapn.), with no evidence found for elimination of mol. cluster fragments. Bond dissocn, energies for cobalt cluster cations, Con+ (n = 2-18), are detd. from measurements of the CID thresholds. Bond energies for neutral cobalt clusters, Co_n (n = 4-18), are derived by combining cationic bond energies with ionization energies for Co_n from literature. The dependence of binding energy on cluster s similar to that obsd. for iron clusters, and inspires some specular regarding cluster ion structures. cobalt clusters, Con (n = 4-18), are derived by combining these cationic bond energies with ionization energies for Con from the literature. The dependence of binding energy on cluster size is

similar to that obsd. for iron clusters, and inspires some speculation

C. A. 1994, 120, NIY

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1994 Con Thao J.J., Han M., et al, J, ceerpyreer, Compret Phess. meep, pacreen Proc. Int Conf., 2nd 1993 (Pub 1994) 296-8. (Ceer. Nig) !!!)

Ho J., Parks E.K., 2hu L., 1995 Con Riley S.J. (25) N=3-16 rhateful a Chera. Phys. 1995, 201 & 1, p 245-261 Reaction of small cobalt clusters With N2: implications for cluster structure.

Con

n=3-13

124: 242739u Theoretical study on electronic structure of cobalt clusters. Li, Jun; Tong, Xiao-min; Li, Jia-ming (Inst. Physics, Academia Sinica, Beijing, Peop. Rep. China 100080). Wuli Xuebao 1995, 44(11), 1727-33 (Ch). Based on multiple-scattering self-consistent-field method, the authors have investigated the electron structure of Conference of Co

meop. pae -: rem 9, compy km. compy km.

cluster (n = 3-13). Comparing the calcd. ionization potentials for different geometries and electron configuration with the exptl. values, the authors can study the appropriate electron configuration and geometric structure of Co_n. From the occupation of local 3d valence electron in configuration, it was found, that the av. at. magnetic moment of Co clusters decreases as the no. of atoms increases. From the geometry of cobalt clusters, the possibilities of various ways of formation of Co_n cluster were also surmised.

C. A. 1996, 124, N 18

1=3,4,6

ronemp.u A. cmp-pa, Opomogn. Cnexmp, Ae

C.A. 1995, 123, NY

123: 41132w Spin-polarized electronic structure of cobalt cluster anions studied by photoelectron spectroscopy. Yoshida, Hiroyuki; Terasaki, Akira; Kobayashi, Katsuyoshi; Tsukada, Masaru; Kondow, Tamotsu (Sch. Sci., Univ. Tokyo, Tokyo, Japan 113). J. Chem. Phys. 1995, 102(15), 5960-5 (Eng). The

photoelectron spectra of Con- (3≤n≤70) were measured at the photon energy of 4.025 eV by use of a XeCl excimer laser. For Con- with n=3, 4, and 6, the geometric and electronic structures were obtained from the spectra in comparison with the calcd. spectra by the spin-polarized DV-Xa method. The spectra obsd. are reproduced reasonably well by the calcn. with postulating the most probable geometrical structures. The 3d band with the majority spin is sepd. by 1.0-2.8 eV from that with the minority spin; the former is completely filled while the latter is partly filled and extends above Fermi level. The magnetic moments and the av. exchange energies of

these cluster anions were estd. For Con- with n≥7, the obsd. electron affinity depends linearly on the reciprocal of the cluster radius and approach the work function of a cobalt metal, as n increases. Below n=6, the electron affinity deviates from the linear dependence. This finding indicates that a size-dependent transition in the electronic structure occurs at n = 7. The spherical conducting drop model suggests the presence of mobile electrons in Con- with n≥7.

Co (kelaconeph) fou &; et al., Yuanzi Yu Fenzi Nuli Xuebao 1996, <u>13</u>(2), 135-140 UK, 46 CREKMIK [Cule Fe (kelacmegn); III)

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Con

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125: 19496n An analytic relationship between size and ionization potential of transition metal clusters. Zhao, Jijun; Chen, Xiaoshuang, Wang, Guanghou (Department of Physics and National Laboratory of Solid State Microstructures, Nanjing University, Nanjing Peop. Rep. China 210093). Chem. Phys. Lett. 1996, 254(1,2), 21-24 (Eng). Based on the effective-coordination-no. (ECN) model, we have constructed an analytic expression for the size dependence of ionization potentials (IPs) for transition metal clusters. An approx. formula of the surface ratio as a function of cluster size for the close-packed structures has been fitted from the icosahedral-like structures. The IPs of Con-Yn, Nbn, and Nin are calcd. and compared with expts. as well as with predictions of the conducting-spherical-droplet (CSD) model. It is found that the present approach agrees with expt. much better than the CSD model.

C. A. 1996, 128, N2

Zhao J., Chen Y., Wang G. 1996 Chew. Phys. Lett., 1996, 254, Con, Yn, Nb,, M/2, p.21-24 9C thrampireck. cooms. Mendy payheriou 4 notery. notheragina Knage pob repex. legasist.

1997 Con Castro M., jamorski Ch., et al. 12 € 5 Cerpykeer., Freeprus Chem. Phys. Lett. 1997, <u>241</u> (1, 2, 3), 133-Cb 934, meop. pacreer. (cue. Fen; !!!)

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F: Con P: 3 Electronic states of transition 131:303468 metal Fujima, Nobuhisa; clusters. Yamaquchi, Tsuvoshi Faculty of Engineering, Shizuoka University Shizuoka 432-8561, Japan Mesosc. Mater. 249-258. Edited by: Arai, Toshihiro. Springer: Berlin, Germany. (English) 1999. A review with 19 refs. We calcd, the electronic states of the 3d transition metal clusters by using the local d.

functional method, and clarified their extraordinary characteristics, which are mainly caused by the chem. bonding 3d electrons in the clusters. We first clarify the structure of chem. bonding in Mn clusters and its size dependence, and discuss the origin of the magic no. which has been obsd. in the cationic clusters. Second, we calc. the electronic states of binary Co-V clusters and discuss the extraordinary stability to the hydrogen-adsorption of Co1 cluster.

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Compostela, Spain). Comput. Mater. Sci. 2001, 22(1-2), 118-122 (Eng). Elsevier Science B.V. Small cobalt clusters Co_n ($2 \le n \le 5$) are studied

by using self-consistent nonlocal generalized gradient approxn. (GGA) d. functional methods. An emphasis is made on a proper treatment of exchange and correlation effects. The enhancement of the magnetic moments as well as the bonding properties of these clusters are discussed in terms of the cluster size and symmetry. We compare some results

of Sambe-Felton and Dunlap (SFD).

D.: Castro, M.; Arias, J. E. (Facultad de Fisica, Departamento de Fisica Aplicada, Universidade de Santiago de Compostela, Santiago de

1.135: 294221r Model potential nonlocal density functional calculations of small cobalt clusters, Con,n ≤ 5. Pereiro, M.; Man'kovsky, S.: Baldomir, D.; Iglesias, M.; Mlynarski, P.; Valladares, M.; Suarez.

from deMon-KS (D) module release 3.2 and the computational scheme

C.A. 2001, 135, N20

F: Co2

P: 3

132:127966 Global potential energy surfaces for the CO2 and CS2 molecules Zuniga, J.; Bastida, A.; Alacid, M.; Requena, A. Departamento de Quimica Fisica, Universidad de Murcia Murcia 30100, Spain Chem. Phys. Lett., 313(3,4), 670-678 (English) 1999 Global potential energy surfaces for the ground electronic states of the CO2 and CS2 mols. are detd. We use the many-body single-value surfaces previously constructed by Murrell and Guo (J. Chem. Soc. Faraday Trans. 2 83. 683) and refine their parameters by the nonlinear leastsquares metho order to reproduce the obsd. vibrational frequencies of each mol. The fi made by computing variationally the vibrational frequencies using sets of optimal generalized internal coordinates. The quality of the potentials checked by computing vibrational frequencies of isotopic species and high excited vibrational energies not included in the fits.

C. A. 2000, 132

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