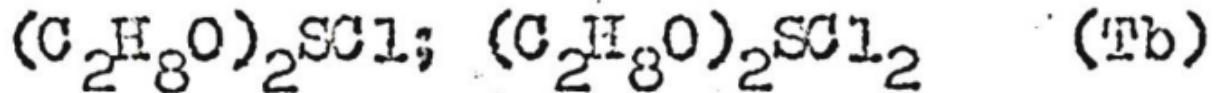
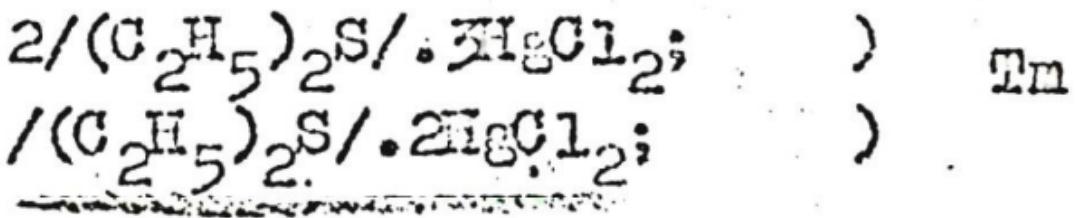


Hg-C-S<sup>2-</sup>deg.

У 1734/1943

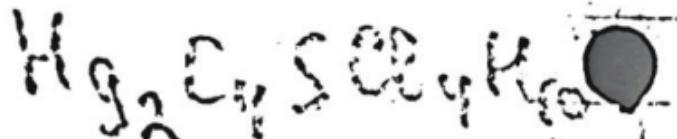


~~Krestov A. E.~~, ~~Kreyzbl A. E.~~

~~J. Russ. Phys.-Chem. Soc.~~; 1929, 61,  
2345-84 ~~Хим. журн. физ.-хим. об-ва~~

The action of zinc...

Be

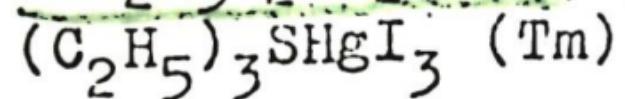
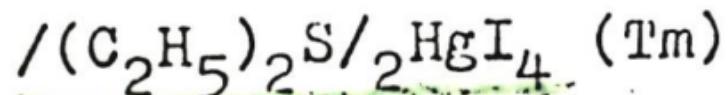


EASTMAN

1

V 1733

1930



Prafulla Chaudra Ray, Nadiabehari Adhikary  
J. Indian Chem. Soc., 1930, 7, 297-303

Complexes of mercuric iodide with alkyl sul-  
fonium iodides.

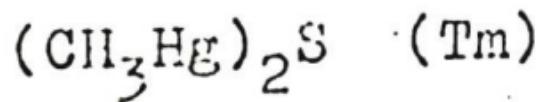
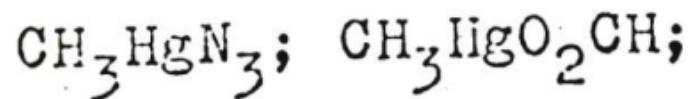
CA, 1930, 4725

Be.

F

V 1727

1933



Perret A., Rerrot R.

Helv., Chim. Acta 1933, 16, 848-57

Some mercuric methyl derivatives.

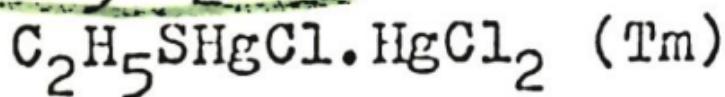
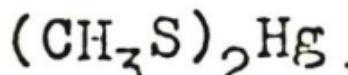
CA., 1933, 4772

Be.

F

V 1725

1938



Blackburn S., Challenger F.  
J. Chem. Soc., 1938, 1872-8

Formation of organo-metalloidal and similar  
compounds by microorganisms. Further studies  
on the fission of the disulfide link.

CA., 1939, 12653

Be.

F

VI 3653 1942

~~5015~~  
HSCH<sub>2</sub>CH<sub>2</sub>SH (Tb), (SCH<sub>2</sub>CH<sub>2</sub>S)Hg,  
(SCH<sub>2</sub>CH<sub>2</sub>S)Pb (T page)

Mathias S.

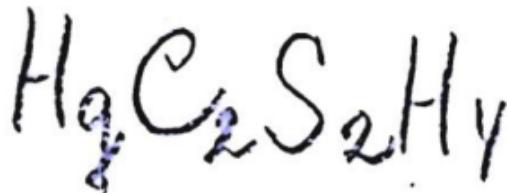
Bols. faculdade filosofia, cienc. letras.

Univ. Sao Paulo 14, Quimica N1, 75-140

1942

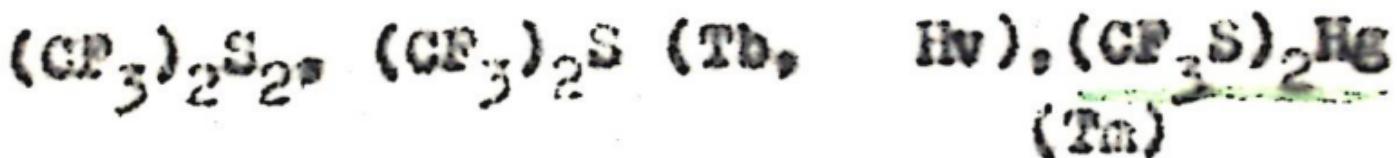
Dimercaptans and...

Be



V 1729 1952

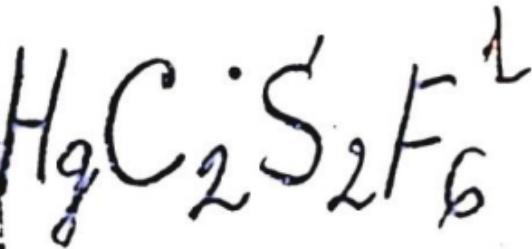
5095



Brandt G.R.A., Emelius H.J., Hasseldine  
R.N.  
J. Chem. Soc., 1952, 2198-205

Organometallic and...

Be



*BP* V 1728

1954

Hg ( $\text{SCH}_2\text{COOH}$ )<sub>2</sub> (K ; H, S)

Stricks W., Kolthoff I.M., Heyndrickx A.

J.Amer. chem. Soc., 1954, 76, N 6, 1515-19

Formation and properties of various mercuric mercapto thioglycolates formed in reactions between mercuric mercury and thioglycolic acid

PJX., 1956, N 2, 3644

Jg.

Est.fotok.

EOTB U. R.

~~CuCl<sub>2</sub>(H<sub>2</sub>O)<sub>2</sub>~~ (89 XVI-2874)

1975

$\text{AgCaM}_2(\text{H}_2\text{O})_2$

$\text{AgCaM}_2(\text{H}_2\text{O})_2$

Kc, NP, Sbf,  
SHf, off;

Saxena R.S., Bha-  
tia S.K.,

J. Indian Chem.

Soc., 1975, 52,  
N9, 785-87

Hg - C -

Hg ( $\text{SCF}_3$ )<sub>2</sub> (Tb, Tm)

| v 1730

1959

Man E.H., Coffman D.D., Muetterties E.L., J.  
Amer. Chem. Soc., 1959, 81, N 14, 3575-77

Synthesis and properties of bis-(trifluoromethylthio) - mercury.

PJX., 1960, 13504

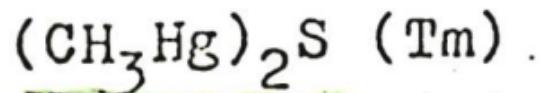
Be.

EOTB 4. 5.

Est. fotok.

V 1726

1958



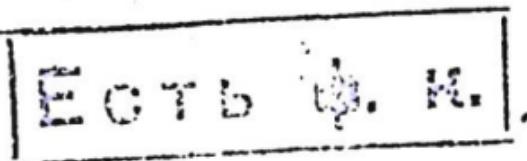
Grdenic D., Markusic B.

J.Chem. Soc., 1958, June, 2434-35

Trismethylmercurisulphonium nitrate and dichro-  
mate.

PJX., 1959, 1345

Be.



Est. Foto.

V 1730

1959

Hg (SCF<sub>3</sub>)<sub>2</sub> (Tb, Tm)

Man E.H., Coffman D.D., Muetterties E.L., J.  
Amer. Chem. Soc., 1959, 81, N 14, 3575-77

Synthesis and properties of bis-(trifluoromethylthio)-mercury.

PJX., 1960, 13504

Be.

EOTB C. H.

Est. fotok.

1961

V 828

$\text{Fe}(\text{Zn}(\text{ClO}_4)_2 \cdot 4(\text{C}_2\text{H}_3)_2\text{SO}, \text{MgCl}_2 \cdot (\text{CH}_3)_2\text{SO},$   
 $\text{CoCl}_2 \cdot 3(\text{CH}_3)_2\text{SO}, \text{NiCl}_2 \cdot 3(\text{CH}_3)_2\text{SO},$   
 $\text{Mn}(\text{ClO}_4)_2 \cdot 6(\text{CH}_3)_2\text{SO}, \text{Co}(\text{ClO}_4)_2 \cdot 6(\text{CH}_3)_2\text{SO},$   
 $\text{FeCl}_3 \cdot 3(\text{CH}_3)_2\text{SO}, \text{CuCl}_2 \cdot (\text{CH}_3)_2\text{SO},$   
 $\text{CuCl}_2 \cdot 2(\text{CH}_3)_2\text{SO}, \text{PdCl}_2 \cdot 2(\text{CH}_3)_2\text{SO}$

Selbin J., Bull W.E., Holmes L.H.  
 J. Inorg. and Nucl. Chem., 1961, 16, N 3-4,  
 219-224

Metallic complexes of dimethylsulphoxide

Aug, 1962, 2377

Bo, Ja E

*1963*  
VI-4111

$Hg_2X_2$  ( $\Delta H_f$ )

X-xanthogenate  $(C_2H_5OCS_2)_2 Hg$

Mukai S., Wakamatsu T.,

J.Mining Metallurg.Inst.Japan, 1963, 79, 12-17

Thermodynamic investigation of reaction  
between mercury and xanthogenate,

M, F. CA, 1966; 64; N1; 75a

DR231

3011-VI  $(CH_3Hg)_2S$ ;  $(CH_3Hg)_2O$   
 $CH_3OHg$ ,  $CH_3SHg$ ,  $CH_3SeHg$  (r<sub>Hg</sub> 1964)  
 $(CH_3Hg)_2Se$  Kesler M.

Croat. Chem. Acta., 1964, 36(3),  
165-8

Structural investigations of ...

J

Phex, 1965, 235261

3133-VI

1965

VI/ $(CF_3Se)_2Hg$ )  $CF_3SHgCl$  (Kf)

Clase H.J., Ebsworth E.A.V.

J.Chem.Soc., 1965, Febr, 940-47

Some reactions and derivatives of bis (tri-fluoromethylseleno) mercury.

PJX., 1965, 24б138

J., M.

Есть оригинал.

3450-VI

1965

$\text{CH}_3\text{HgF}$ ,  $\text{CH}_3\text{HgCl}$ ,  $\text{CH}_3\text{HgBr}$ ,  $\text{CH}_3\text{HgJ}$ ,  $\text{CH}_3\text{HgOH}$ ,  
 $(\text{CH}_3\text{Hg})_2\text{HPO}_4$ ,  $(\text{CH}_3\text{Hg})_2\text{S}$ ,  $\text{CH}_3\text{HgS}^-$ ,  $\text{CH}_3\text{HgSCN}$  (Kp)

Schwarzenbach G., Schellenberg M.

Helv. chim. acta, 1965, 48, N 1, 28-46

Die Komplexchemie des Methylquecusiüber-Kations

PJX, 1965, 19B54

Ja.

Есть оригинал orig.

Hg - C - KOMPLEKSE

Bp - 4494-VI 1966

Meric complexes in dimethylformamide. Mylene Breant and Nguyen-Van-Kiet (Inst. Natl. Scis. Appl., Villeurbanne, France). *Compt. Rend.*, Ser. C 262(12), 955-8(1966)(Fr). A study of the polarographic waves during the oxidn. of Hg in the presence of  $\text{Cl}^-$ ,  $\text{Br}^-$ ,  $\text{I}^-$ , thiourea,  $\text{NCS}^-$ , and thioglycolic acid in  $\text{HCONMe}_2$  revealed the presence of complexes of  $\text{Hg}^{++}$ . Analysis of the waves and a study of their displacement led to the detn. of the formulas and the dissocn. consts. of the complexes. At an ionic strength of 0.1, the following pK values were found:  $\text{Hg}(\text{HSCH}_2\text{COO})_2$ , 39.6;  $\text{HgI}_4^{2-}$ , 36.6;  $\text{HgBr}_4^{2-}$ , 35.2;  $\text{HgCl}_4^{2-}$ , 22.5;  $\text{Hg}[\text{SC}(\text{NH}_2)_2]_4^{2+}$ , 26.5; and  $\text{Hg}(\text{SCN})_4^{2-}$ , 22.5. The pK of thioglycolic acid in dimethylformamide at ionic strength 0.1 is  $9.25 \pm 0.1$ .

D. H. Hutson

C.A. 1966, 65:2

1461ef

+1



$(CH_3)_3SOHgJ_3$  ( $T_m$ ) V14759 1967

$(CH_3)_3S^+$  (an. noef.) 5, 10 (cp)

Greighton J.A., Green J.H., Harrison D.J.  
Waller S.M.

Spectroclim. Acta, Part A, 1967, 23(12),

The vibrational  
triethylsulfonium  
xonium compounds.

specbre of some  
and triethylsulfo  
CS, 1968, 68, 16, 2522Kw  
2973-9

S(CH<sub>2</sub>CH<sub>2</sub>OH)<sub>2</sub>Hg

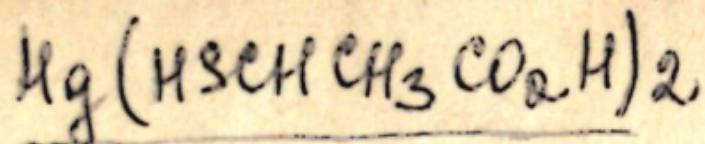
1941

(Ketab.)

91921p Thermodynamics of the formation of  $\beta$ -thiodiglycol complexes with mercury(II) ion. Mogilyanskii, A. I.; Sukhova,

Hg[S(CH<sub>2</sub>CH<sub>2</sub>OH)<sub>2</sub>] T. G.; Temkin, O. N.; Flid, R. M. (Mosk. Inst. Tonkoi Khim. Tekhnol. im. Lomonosova, Moscow, USSR). *Zh. Neorg. Khim.* 1971, 16(7), 2017-18 (Russ). The stability consts. are given for complexes of S(CH<sub>2</sub>CH<sub>2</sub>OH)<sub>2</sub> (L) with Hg<sup>2+</sup> (HgL<sub>1</sub><sup>2+</sup>, HG<sub>2</sub>L<sub>1</sub><sup>2+</sup>, HG<sub>3</sub>L<sub>1</sub><sup>2+</sup> and HG<sub>4</sub>L<sub>1</sub><sup>2+</sup>). The stability of HG<sub>2</sub>L<sub>1</sub><sup>2+</sup> and HG<sub>3</sub>L<sub>1</sub><sup>2+</sup> explains the effect of small concns. (0.3-2 wt. %) of L on catalytic activity of the HgSO<sub>4</sub>-Fe<sub>2</sub>(SO<sub>4</sub>)<sub>3</sub>-H<sub>2</sub>SO<sub>4</sub> system in hydration of C<sub>2</sub>H<sub>6</sub> (1963).

C.A. 1941. 5. 14



1972

Koushik

(Kras)

Srivastava S.K.

Srivastava P.C

Nigam H.L.

"Indian J. Chem" 1972,

10(2), 232-4 (eng)

T<sub>KAFT</sub> (CuCl-H-S-e-O)

$\text{Mg}(\text{SCN})_2$  (aq.)      Omnilek 14725 } 1980

Collected near B. H., Teng-  
zohor B. d.

Kp. Srb;

Zh. Zool. Zemph., 1980,  
46, N 11, 1129-1134.