

96 Ben

Wartenberg and Bosse (9) 1922 ^{3 3-11}

Volmer (2) 1923

Geeher and Jellinek (3) 1933

Tahn-Held and Jellinek (4) 1937

Bloom et al 1958

Margulescu and Maria 1961

Bloom and Hastie 1968.

omge $D_0 = 5,42$ et ge $P_0 P_{02}(z) = P_0(z) + 2B_{2/2}$

Q/B22

JANAF

1975.

J. Phys Chem. Ref Data
Vol 4, N1.

Chase M. W. et al.

$$\Delta H_f^0 = -21,03 \pm 1,5 \text{ kcal/mol}$$

$$\Delta H_{298,15} = -24,95 \pm 1,5 \text{ kcal}$$

but also to Π and Π 3-mer as

gammox more  7 per cent

$$\Delta H_{298} = 33,28 \text{ kcal.}$$

Wartenberg and Bosse (1) 1922 ³³⁻¹¹

Volmer (2) 1923

Geeher and Jellinek (3) 1933

Tahn-Held and Jellinek (4) 1937

Bloom et al 1958

Muzulescu and Maria 1961

Bloom and Hastie 1968.

Omge $D_0^0 = 5,42 \text{ eV}$ and $P(B_2(z)) = P(z) + 2B_2(z)$

96B22

XIV-5577

1973

(Mj) Murgulescu I. G. et al.

Rev. Roum. Chim., 1973,
18, No, 1667-80

961322

ΔH_f

1721

1963

Brewer L.

J. Chem. Rev. 1963, 63, 111

5808-X

1968

PbBr₂ ?

Bloom H. Hastie W.

$\Delta_f H = 38.3 \pm 1.0$
kcal/m

Austral J. Chem. 1968, 21, (3)

583-587

"Vapour phase composition and
activities for the molten
PbBr₂ + KBr system"

Pl Br

V-773

1958

(b) Bloom H., Bockris J. O. M.

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No, 2044-46

ВФ-12058 IV

1934

РБ РЗ₂

Д₅Н.

Д₅Н.

Понов Б,

ЗН. физ. хим. 1934, 5, 863-84.

PbBr₂ + X-3415 1933

(p) Greiner B., Fellinek K.

Z. Phys. Chem. 1933, A165,
N3, 97-120