

D



Johnston, H L, 1934

D Long E. A,
A. Chem Phys 2, 389 (1934)

39p-5005-1

Johnston H. L., 1934

D.

Long, E. A.:

y. Chem Phys 2, 710/1934)

Д

Zeise

1954

Термодинамика

т-д
ф-цм
до 3000

Ссылка у Zeise-[446]- на работу

1962

Д
за

Гурьев Л. В. и др.

Москва, 1962

т. 1.

Периодические
сва и другие
и в.

D
rag

1965

Wagman A.D. et al
NBS, Tech. Note 240-1, Oct 1965
Washington

Selected Values of Chemical Thermo-
dynamic Properties Part I

$$H_{298}^{\circ} - H_0^{\circ} = 1481 \text{ kcal/mole}$$

$$S_{298}^{\circ} = 29,455 \text{ kcal/mole grad}$$

$$C_{298}^{\circ} = 4,9679 \text{ kcal/mole grad}$$

D

Think T.P.
at el

1971

(ug. rag)

Hydrocarbon Process
1971, 50(1), 98-104

cp*

Cell C, I

E.H. 1971. 78. 14

$\delta^-(z)$

1977

YANAF

m.g.p.

0-6000

March 31, 1977.



$\Phi^+(z)$

1977

YANAF

m.g.op.

0-6000

March 31, 1977 .



D (2)

1974

JANAF

m.g. q

0-6000

March 31, 1974 .

D

1980

m. gum.

gs - gum

0-6000°K.

(+26) ☒C.A. 1980. 93 N12

93: 102312r Ideal gas thermodynamic functions of hydrides and deuterides. Part I. Burcat, Alexander (Dep. Aeronaut. Eng., Technion-Israel Inst. Technol., Haifa, Israel). TAE Rep. 1980, TAE No. 411, 80 pp. (Eng). Thermodyn. data at 0-6000 K are tabulated for the species D, H, D₂, H₂, HD, CD, CH, CD₂, CH₂, CD₃, CH₃, CD₄, CD₃H, CD₂H₂, CDH₃, CH₄, C₂D₄, C₂H₄, C₂D₆, C₂H₆, OD, OH, H₂O, D₂O, HDO, CDO, and CHO. All the data are in SI units. Equations representing the properties are given.

D (g)

1985

YANAF

T. p.

II 439. 1985 стр. 999

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