86-Hal

Sb-Hal

BD-5036-IV	1935

Hal=Cl, J

Butkov K,

спектр

Acta physicochim. URSS,
1935, III, N2-3, 205-218

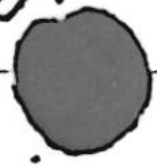
1967

SB Hal

Kolditz L.

Halogen Chem. V. 2., London - New York, Acad. Press, 1967, 115-168

Галогениды металлов и сурьмы



☒

SbHal$_6^{3-}$ | X = Cl, Br, I. 1970

(58910r) Vibrational spectra of SbX_6^{3-} and TeX_6^{2-} anions: new observations on the singular properties of some systems related to XeF_6. Adams, Christopher John; Downs, A. John (Dep. Inorg. Chem., Univ. Oxford, Oxford, Engl.). *J. Chem. Soc. D* 1970, (24), 1699–701 (Eng). Anal. of Raman and far-ir spectra indicates that TeX_6^{2-} and SbX_6^{3-} differ from conventional octahedral systems: TeX_6^{2-} has an octahedral ground state and less symmetrical electronically excited states; SbX_6^{3-} has a distorted octahedral configuration with C_{3v} symmetry (as does the instantaneous configuration of XeF_7) and is very sensitive to environment. For TeX_6^{2-}, the ratio of intensities of ν_1 and ν_2 vibrations is much larger (~10 fold) than for octahedral species; the ν_3 vibration is clearly perceptible in the Raman spectrum; no feature attributable to ν_4 was detected at >70 cm^{-1}. In the region assocd. with Sb-X stretching modes, 3 Raman lines, 2 of which were polarized and coincident with ir absorptions, were obsd. FBJN

Hal = Cl, Br, I

Di, стр-ра

(+1)

C.A. 1971. 74. 12.

SbX_6 n-

X = F, Cl, Br

n = 1, 2

Cubic n.

Rao D.V.R.A. et al. 1970

Proc. Indian Acad. Sci., A71 (1), 42

(Calc. P-Hal) III

70225.461[illegible]7
Ph, Ch, TC

X = Cl, Br, J.
34457.
SbX_6^{3-} (сил. пост. ср. амплит. колеб.)

1976
№ 4-17079

Sa[illegible]nyal Nitish K., Verma D.N., Dixit L. Force fields and vibrational mean amplitudes of SbX_6^{3-} and BiX_6^{3-}, X=Cl, Br, and I. "Indian J. Phys.", 1976, 50, N 4, 483-487 (англ.)

0822 ПМК

777 777

8 (1)

ВИНИТИ

$SbCl_6^{3-}$; $SbBr_6^{3-}$; SbI_6^{3-} 1976

85: 151062c Force fields and vibrational mean amplitudes of SbX_6^{-3} and BiX_6^{-3}, X = chlorine, bromine and iodine. Sanyal, Nitish K.; Verma, D. N.; Dixit, L. (Dep. Phys., Univ. Gorakhpur, Gorakhpur, India). *Indian J. Phys.* 1976, 50(4), 483–7 (Eng). A normal coordinate anal. of SbX_6^{3-} and BiX_6^{3-} where X = Cl, Br, and I, is presented using the *F–G* matrix method of E. B. Wilson (1955) using GVFF and MUBFF potential functions. Trends in force constants are discussed and mean amplitudes of vibration for principal distances in octahedral systems are presented to supplement the force field data. The characteristic mean amplitudes of vibration in isostructural ions were also examd.

cal, vcf.

(+5)

C.A. 1976. 85, N20

SbX_3 1977

X = галоген.

Hyde Robert G.,
Peel J. Barrie.

„Aust. J. Phys.", 1977, 33, № 3, 887-896.

расчет
электр.
строен.

(см. SnX_4) III

Sb X5

1979

Х-лазер

Drake M.C., Rosenblatt G.M.,
Raman spectroscopy in high
temperature chemistry.

СКР

10th Materials Research Sympo-
sium on characterization of
High temperature, Vapors and
Gases.
NBS Special Publication 561.
Volume I, 1979, 609–646.
(у Гурвича)

Sб Х3
Х-галоген

1979

Drake M.C., Rosenblatt G.M.,
Raman spectroscopy in high temperature chemistry.
10th Materials Research Symposium on characterization of High temperature, Vapors and Gases.
NBS Special Publication 561
Volume 1, 1979, 609-646.

СКР

● (у Гурвича)

SbX_6^{3-}

SbX_5^{2-}

$Sb_2X_9^{3-}$

SbX_4^-

X = Br, I.

(Ic)

1980

94: 9404h Low-frequency vibrational spectra of bromo- and iodoantimonates. Jagodzinski, Paul W.; Laane, Jaan (Dep. Chem., Texas A and M Univ., College Station, TX 77843 USA). *J. Raman Spectrosc.* 1980, 9(1), 22–7 (Eng). The Raman and far-IR spectra of antimony halide complexes of formula SbX_6^{3-}, SbX_5^{2-}, $Sb_2X_9^{3-}$ and SbX_4^{1-} were recorded and analyzed for X = Br and I. The stretching frequencies for the bromo complexes range from ~ 135 to 220 cm^{-1} for external bonds and from ~ 110 to 130 cm^{-1} for bridging bonds. For the iodo complexes, the ranges obsd. are 138–174 cm^{-1} (external) and 87–127 cm^{-1} (bridging). Approx. force consts. were calcd. for the anions, and these were compared to values previously detd. for halobismuthates. In general, the ordering of the force consts. was Sb–Br > Sr–I ~ Bi–Br > Bi–I. As was found for the Bi complexes, the stretching force consts. were greatest for bonds in the anions in which the largest no. of other bonds were to bridging halogens.

C.A. 1981, 94:12

1980

SbX_6^-

(см. пост. м.л.)

X = галоген

(см. LiX_6^{--}; III)

Sarkar P.C., et al.,

Indian J. Pure Appl. Phys. 1980, 18 (7), 516-23

SbY$_6$ 1982

Y-галоген

Mohan S., Mukunthan A.

см. посл.

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(см. SeY$_6$; III)

SbH_3-HX | Ом-21276 | 1984

X=F, Cl

Arlinghaus R. T.,
Andrews L.,

ИК спектр в матрице.

J. Chem. Phys., 1984, 81, N 10, 4341-4351.

$Sb X_3$ 1984

Х-рентген. Grodzicki Michael, Walther Horst, et al.

фото- Z. Naturforsch., 1984,

электрон. B 39, N 10, 1319-1330.

спектры

(см. $N X_3$; III)

SbX_3

$X = F, Cl, Br.$

1984

Xin Min, Chow Chiu
Lee-Yung, et al.

J. Electron. Spectrosc.
and Relat. Phenom.,
1984, 33, N2, 93–105.

Электронная структура.

(см. NX_3; III)

Галогениды Sb 1986

Кузнецов С. Л.,
Харитонов Ю. Я.

Ж. неорган. химии,
1986, 31, N 7, 1716-1722

(см. Галогениды P, III)

Галогениды Sb 1986

Kuznetsov S. L.,
Vi, структ. Kharitonov Yu. Ya.
параам., Zh. Neorg. Khim.
сеел. 1986, 31 (7), 1716-22.
по сей.

(сеи. Галогениды P; III)

SbX_3 (от. 31 177) 1988

X=F, Cl, Br, Sakai Y., Miyoshi E.,
g

теоретр., M, теорет. расчет.

J. Chem. Phys. 1988, 89, N 7, 4452–4453.

$[SbX_6]^{n-}$

1991

$n = 1, 2, 3$

X - галоген

116: 70739k Calculation of force constants and vibrational frequencies of octahedral metal halide (MX_6) molecules. Hahn, Egbert; Hebisch, Ralph (Anal. Cent., Cent. Inst. Phys. Chem., 1199 Berlin, Germany). *Spectrochim. Acta, Part A* 1991, 47A(8), 1097–101 (Eng). The vibrational frequencies of some octahedral species ($[SbX_6]^-$, $[SbX_6]^{2-}$, $[NbX_6]^-$, $[NbX_6]^{2-}$, $[TaX_6]^-$, $[TaX_6]^{2-}$; X = F, Cl, Br, I) were calcd. by means of 6 extrapolated mol. force consts. using some linear relations between the force consts. and the reciprocal radii of the ligands. A statistical treatment of these correlations allowed the calcn. of error limits for a probability of 90%. The computations of the force consts. and vibrational frequencies were based on the GF-matrix method.

Coul. 20CM.

(42)

C. A. 1992, 116, N8

SbX3
SbX5
X=F÷P

1992

Breiding J., Thiel W.,

J. Comput. Chem. 1992,

13 (2), 165-76

ab initio

расчет

(см. PX3, PX5 X=F÷P)

86[3

F=F, Cl, Br, I

ОП 36939

1992

Егоров Ю.С.,

ж. физ. химии, 1992, 66, N 12, 3258–3263.

сил. пост. структура

Силовые постоянные, константы Кориолисова взаимодействия и особенности

строения тригалогенидов
фосфора, мышьяка и сурьмы.