

CF<sub>2</sub>

1995

23 Б1223. Инфракрасная лазерная спектроскопия полосы  $\nu_2$  дифторкарбена (CF<sub>2</sub>). Infrared laser spectroscopy of the  $\nu_2$  band of difluorocarbene (CF<sub>2</sub>) / Qian Hai-bo, Davies Paul B. // J. Mol. Spectrosc. — 1995. — 169, № 1. — С. 201—210. — Англ.

В области 638—706 см<sup>-1</sup> с разрешением, ограниченным доплеровским уширением, измерена (полупроводниковый лазерный спектрометр) вращательная структура полосы  $\nu_2$  молекулы CF<sub>2</sub>( $\tilde{X}^1A_1$ ). Приведено положение и отнесение 217 наблюдаемых линий ( $J \leq 40, K_a \leq 7$ ). При анализе спектра использован А-приведенный гамильтониан Ватсона в I'-представлении. Значения молекулярных постоянных для основного состояния фиксировали в соответствии с данными исследований микроволновых спектров. Начало полосы при 666,24922 см<sup>-1</sup>. Значения вращательных постоянных А, В и С CF<sub>2</sub> ( $\nu_2=1$ ) (в см<sup>-1</sup>): 2,97347; 0,416141; 0,363055. Приведены также значения параметров центробежного искажения.

В. М. Ковба

М.А.

X. 1995, N 23

CF<sub>2</sub>

DM-38120

1995

Roberto Martinez, Inaki  
Merelas et al.,

( $\tilde{A}-\tilde{X}$ )

J. Phys. B: At. Mol.  
Opt. Phys. 1995, 28,  
4563-4575

CF<sub>2</sub>

1995

Su, Lohi; Deakynne, C.A.,  
et al.

(g) Nali ~~Huaxue~~ Xuebao  
1995, 11 (10), 885-9

(coll. Bet, I)

CF2

(im 38994)

1997

монохром.  
сп-ра,  
комплан.  
расстотн,  
ДФН

Byeong-keo Cheong<sup>†</sup>,  
Kun - book Cho,

J. Phys. Chem., A 1997,  
101, 7901-7906

CF2

Om. 39136

1997

Josefundo R. Pliers Jr et al.,

J. Chem. Soc., Faraday Trans.,

1997, 93(10), 1881-1883.

Ap,  
at initio  
paint

Absolute  
affinity and



proton  
basicity of

the carbonates  $\text{CH}_2$ ,  $\text{CF}_2$ ,  $\text{CCl}_2$ ,  
 $\text{C}(\text{OH})_2$ ,  $\text{FCOH}$ ,  $\text{CPh}_2$  and  
fluorenylidene.

1997

CF<sub>2</sub>

127: 41511b Determination of A-state stretching frequencies of CF<sub>2</sub> radical. Wang, Chuji; Xiao, Yingsheng; Ran, Qin; Chu, Yiwei; Chen, Congxing; Yu, Shuqin; Ma, Xingxiao (Dep. Chem. Phys., Univ. Sci. Technol. China, Hefei, Peop. Rep. China 230026). *Wuli Huaxue Xuebao* 1997, 13(4), 297-300 (Ch), Beijing Daxue Chubanshe. Laser induced fluorescence excitation spectroscopy was used to det. the previously ambiguous assignments of A-state stretching frequencies of CF<sub>2</sub> radical under supersonic free jet conditions. The measured frequencies are  $\nu_1 = 1012.1 \pm 0.5 \text{ cm}^{-1}$ ,  $\nu_3 = 1180.2 \pm 0.5 \text{ cm}^{-1}$ , which is in good agreement with Cameron's<sup>(6)</sup> calcn. result. Also, some transitions attributed to the (1,n-2,0)-(0,0,0), n ≤ 6 progressions are 1st reported, and new parameters are derived from the spectra obtained.

 $\nu_1$  &  $\tilde{\nu}_{\text{COem}}$ 

C. A. 1997, 127, N 3

CF<sub>2</sub>

1998

129: 46910g Laser-induced fluorescence studies of jet-cooled CF<sub>2</sub>: determination of A-state stretching frequencies. Wang, Chuji; Chen, Congxiang; Dai, Jinghua; Ma, Xingxiao (Department of Chemical Physics, University of Science and Technology of China, Hefei, Peop. Rep. China). *Chem. Phys. Lett.* 1998, 288(2,3,4), 473-480 (Eng), Elsevier Science B.V.. Laser-induced fluorescence (LIF) excitation spectra of the CF<sub>2</sub> A <sup>1</sup>B<sub>1</sub>-X <sup>1</sup>A<sub>1</sub> system around 250 nm was measured under supersonic free-jet conditions. CF<sub>2</sub> radical was produced by DC-discharge of CF<sub>4</sub> seeded in Ar. Previously ambiguous assignments of the A-state stretching frequencies were exptl. detd. to be  $\nu_1'=1012.1$  cm<sup>-1</sup> and  $\nu_3'=1180.2$  cm<sup>-1</sup>, by analyzing the subband spectra attributed to some of the obsd. progressions (2<sub>0</sub><sup>n</sup>, 1<sub>0</sub><sup>1</sup>2<sub>0</sub><sup>n-2</sup> and 2<sub>0</sub><sup>n</sup>3<sub>1</sub><sup>1</sup> (n ≤ 6)) in which 21 vibronic bands were unambiguously labeled. The measured A-state stretching frequencies do not support King et al.'s assignments [D.S. King, P.K. Schenck, J.C. Stephenson, *J. Mol. spectrom.* 78(1979) 1], but are in good agreement with Cameron et al.'s theor. assertions [M.R. Cameron, S.H. Kable, G.B. Bacskay, *J. Chem. phys.*, 103(1995) 4475].

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флуорес.

A' B<sub>1</sub> - X' A<sub>1</sub>

Di

C.A. 1998, 129, N 4



F: CF2

P: 3

131:134914 Structure and Energetics of Group 14 (IV-A) Halides: A Comparison Density Functional-Pseudopotential Study. Escalante, Sigfrido; Vargas, Rubicelia; Vela, Alberto (Departamento de Quimica Inorganica Posgrado Fac de Quimica, Universidad Nacional Autonoma de Mexico, Mexico D.F. 04510, M J. Phys. Chem. A, 103(28), 5590-5601 (English) The complete set of MX<sub>2</sub> and MX<sub>4</sub> (M = C, Si, Ge, Sn, Pb and X = F, Cl, Br, group 14 halides are studied with d. functional theory and quasirelativistic effective core potentials. To analyze the role of d. inhomogeneities and asymptotic behavior of the Kohn-Sham effective potential in these molecules, following exchange-correlation energy functionals are tested: local, semi (generalized gradient), and hybrid functionals. For comparison, Hartree-

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C.A. 1999, 131

1999

results are also presented. Fully optimized geometries are in very good agreement with exptl. available data and with other high-level theor. cal. The energy differences assocd. with the dissocn. and disproportionation reactions are reported. Zero-point corrections and at. spin-orbit effects included in these reaction energies. The dissocn. energies predicted at Hartree-Fock level are underestimated, the local energy differences are overestimated, and both the semilocal and hybrid approaches provide the b ests. for these reaction energies. The disproportionation energies, whic commonly used to explore the relative stability of different at. valences a behavior that departs from that commonly known for reactions involving single atom: the local and semilocal disproportionation energies have ver similar values and follow the same trends.

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F: CF2

P: 3

130:305739 Measurement of CF3 and CF2 radicals in a hollow cathode CF4 plasma. Fujioka, Hiroyuki; Mashino, Souichi; Aikyo, Satoshi; Goto, Miki; Arai, Toshihiko (Department of Electrical Engineering, Kanagawa Institute Technology, Japan).

Kanagawa Koka Daigaku Kenkyu Hokoku, B, B-23, 69-72 (Japanese) 1999 Neutral CF3 and CF2 radicals in a hollow cathode dc CF4 discharge have been first measured under various discharge currents and CF4 pressures with quadrupole mass spectrometer, using the threshold ionization technique. It has been found that the CF3 and CF2 radical densities were higher than those obtained for other types of discharges.

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CF<sub>2</sub>

DM. 401071

1999

ab initio  
geom.,  
konform.  
zamenom,  
 $\Delta_f H$  (pauze)

Martin Schwartz<sup>†</sup> and  
Paul Marshall<sup>†</sup>

J. Phys. Chem. A 1999,  
103, 7900-7906



$\text{CF}_2^-$

[Am. 40 099]

1999

Rebecca L. Schwartz,  
Gustavo E. Davico et al.,

J. Phys. Chem., 1999, A103,

8213 - 8221.

Singlet - Triplet splittings  
in  $\text{CX}_2$  ( $X = \text{F}, \text{Cl}, \text{Br}, \text{I}$ )

Dihalocarbenes via Negative  
Ion Photoelectron Spectroscopy



2000

F: CF2

P: 3

133:302781 Nuclear spin-rotation interaction in CF2 (X 1A1) observed by Fourier transform microwave spectroscopy. Hansen, N.; Mader, H.; Temps, F.

Institut für Physikalische Chemie,  
Christian-Albrechts-Universität zu Kiel Kiel  
D-24098, Germany Chem. Phys. Lett., 327(1,2),

97-103 (English) 2000. The rotational spectrum of difluoromethylene (CF2) in its X 1A1 ground electronic state was obsd. using mol. beam-Fourier transform microwave spectroscopy. CF2 was produced in a free jet expansion in Ar as inert



carrier gas by ArF excimer laser photolysis of  $\text{CF}_2\text{Cl}_2$ ,  $\text{CF}_2\text{CHCl}$ , or  $\text{CF}_2\text{CHBr}$  or by flash pyrolysis of  $\text{CF}_2\text{Br}_2$  in the high-pressure region directly behind a pulsed mol. beam valve. Rotational transitions were obsd. in the frequency range from 8 to 40 GHz in the  $\text{CF}_2$  (000) ground vibrational state and in the (100), (010), and (001) excited vibrational states. Several obsd. lines were split due to the hyperfine interaction between the  $^{19}\text{F}$  nuclear spin and the overall rotation of the mol. The results were used to det. the nuclear spin-rotation interaction parameters for  $\text{CF}_2$ .

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F: CF2

P: 3

2000

132:214053 Spectroscopic constants of the X(1A1), a(3B1), and A(1B1) stat of CF2, CCl2, and CBr2 and heats of formation of selected halocarbenes: a initio quantum chemical study.

Sendt, Karina; Bacskay, George B. School of Chemistry, University of Sydney Sydney 2006, Australia J. Chem. Phys., 112(5), 2227-2238 (English) 2000 The geometries, rotational const., harmonic force const. and frequencies, dissocn. and term energies of CF2, CCl2, and CBr2 in their r X(1A1), a(3B1) and A(1B1) states, computed by complete active space SCF (CASSCF), complete active space 2nd-order perturbation (CASPT2), and coup cluster with single, double and perturbative triple excitations [CCSD(T)] methods and cc-pVTZ basis sets, are reported. For CCl2 and CBr2 the barr to linearity are characterized. The computed

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CA 2000, 132

spectroscopic consts. are in good agreement with the available exptl. data. The atomization energies hence heats of formation at 0 and 298 K of these mols. and of CHF, CHCl, CFCl, all in their lowest singlet ground states were computed by the CCSD method using basis sets ranging from cc-pVDZ to aug-cc-pVQZ, cc-pCVQZ and G3large, enabling the extrapolation of the energies to a complete basis s (CBS) limit and the inclusion of core-valence correlation (with the excep of CBr2). Scalar relativistic corrections, computed by CASPT2, were take into account. The final CBS results (in kJ mol-1) are .DELTA.fh2980(CHF) 147 .+- . 4, .DELTA.fh2980(CHCl) = 320 .+- . 4, .DELTA.fh2980(CFCl) = 29 .+ 4, .DELTA.fh2980(CCl2) = 229 .+- . 4, and .DELTA.fh2980(CBr2) = 337 .+- . 8

The GAUSSIAN-3 (G3) heats of formation of these mols. are within 2 kJ mol of the CBS values, representing significant improvement over the GAUSSIAN predictions which differ from the CBS results by up to ~10 kJ mol-1. Similarly, the G3 triplet/singlet energy sepns. are in agreement with tho obtained at the CCSD(T)/cc-pVQZ level of theory including core-valence correlation corrections, as well as expt.