

Re H, Re H₂

ReH

1974

Preston H. J. T., et al.

(Y; Ae)

Int. J. Quant. Chem.,
1974, 12(3), 471-84

● (see. Mo; 14)

Ref⁺

[Om 34017]

1990

Charessian G., Breusich et al.
et al.,

neop.
pocrem

J. Amer. Chem. Soc. 1990,
112, 7179-7189.

Theoretical study of Transition-
Metal Hydrides.

5. HfH^+ through HgH^+ ,
 BaH^+ , and LaH^+ .

ReH_2
 ReH_2^+

1991

4 Д177. Поверхности потенциальной энергии для внедрения Re и Re^+ в H_2 . Potential energy surfaces for the insertion of Re and Re^+ into H_2 / Dai Dingguo, Balasubramanian K. // J. Chem. Phys.— 1991.— 95, № 6.— С. 4284—4295.— Англ.

Рассчитаны поверхности потенц. энергий (ППЭ) 12 электронных состояний ReH_2 и 13 электронных состояний ReH_2^+ . Применялся метод ПАП—МКССП с включением $1,5 \cdot 10^6$ конфигураций. Получено, что в основных состояниях $\text{Re}(^6\text{S})$ и $\text{Re}^+(^7\text{S})$ не внедряются в H_2 , в то время как возбужденные $\text{Re}(^6\text{D})$ и $\text{Re}^+(^5\text{G})$ спонтанно внедряются в H_2 . Основное состояние ReH_2 высокоспиновое линейное $^6\Sigma_g^+$, а основное состояние ReH_2^+ — изогнутое $^5\text{B}_2$ -состояние. Рассмотрены также некоторые низколежащие возбужденные состояния. Для ReH_2 и ReH_2^+ проявляются спин-орбитальные эффекты. Возбужденные состояния $\text{ReH}_2^+(^3\text{B}_2)$ и $(^1\text{B}_2)$ имеют ППЭ с двумя минимумами.

М.П.

ср. 1992, № 4

ReH₂

1992

118: 132505h Theoretical study* of rhenium hydride (ReH₂). Swang, O.; Faegri, K.; Gropen, O. (Dep. Chem., Univ. Oslo, N-0315 Oslo, Norway). NATO ASI Ser., Ser. B 1992, 283(Cluster Models Surf. Bulk Phenom.), 463-72 (Eng). The potential energy surfaces for a no. of quartet and sextet states of the ReH₂ were studied using ab initio CASSCF-CI and MCPF calcns. Different basis sets were applied; the treatment of correlation effects is discussed. The global energy min. of the ReH₂ is obtained with the H₂ mol. infinitely sepd. from the Re atom in its ⁶S state. The most stable bound state is linear ⁶Σ_g⁺ with R(Re-H) = 1.88 Å 12.2 kcal/mol above the ref. state. A ⁴B₁ state with R(Re-H) = 1.72 Å and a bond angle of 117 degrees is found 19.8 kcal/mol above the ref. These two energy min. are analogous to these of the H-Re-CH₃ system. The relationship between bonding energies and at. spectra is discussed.

meop. parum

C.A. 1993, 118, N14

ReH

1993

118: 179024r Spectroscopic properties and potential energy curves for 30 electronic states of rhenium monohydride. Dai, Dingguo; Balasubramanian, K. (Dep. Chem., Arizona State Univ., Tempe, AZ 85287-1604 USA). *J. Mol. Spectrosc.* 1993, 156(2), 455-67 (Eng). Spectroscopic consts. and potential energy curves of 30 electronic states of the diat. ReH are computed using complete active space multiconfiguration SCF followed by first-order CI and multireference singles + doubles CI calcs. The second-order CI calcs. are made on four low-lying electronic states ($T_e < 10,000$ cm^{-1}). The authors also carry out relativistic CI calcs. including spin-orbit coupling to compute spin-orbit effects on the low-lying states of ReH. The ground state of ReH is found to be of $^7\Sigma^+$ symmetry ($R_e = 1.82$ Å, $\omega_e = 1611$ cm^{-1} , $D_e = 1.3$ eV) with a very low-lying excited state of $^3\Sigma^+$ symmetry ($T_e \sim 3000$ cm^{-1}) in the absence of spin-orbit effects. The anal. of the wavefunctions and Mulliken populations reveals that all the low-lying electronic states of ReH are very ionic (Re^+H^-). The Re atom has an effective configuration of $6s^{1.15}6p^{0.45}5d^{3.01}$ and $6s^{1.75}6p^{0.06}5d^{3.09}$ in the $^7\Sigma^+$ and $^3\Sigma^+$ states, resp. The Re-H σ bond in the $^7\Sigma^+$ ground state has 45% Re(s), 16% Re(p), and 39% Re(d) character while in the $^3\Sigma^+$ state, it has 60% Re(s), 2% Re(p), and 38% Re(d) character.

M.N., NOMER
Q-uc

C.A. 1993, 118, N18

Re H₂ Balasubramanian K. 1994

Adv. Met. Semicond.

Clusters 1994, 2,
115-36.

структурн.
параметры,
сил. пост.
эл. сост., скин-
орбит. взаи-
моу., теор.
расчет.

(ср.  H_fH₂; III)