

$\text{PtCl}_2$

VI-53

1961

VI (PtCl<sub>2</sub>, K<sub>2</sub>PtCl<sub>4</sub>; K<sub>2</sub>XCl<sub>6</sub>;  
(X=Re, Os, Jr. Pt, Pd)

Adams D.M.

Proc. Chem. Soc., 1961, Sept, 335.

Infrared-active metal-chlorine stretching  
frequencies.

RX, 1962, 5B102.

Est/EET b Q. M.

J.

3086-VI

1963

Wi ( $\text{CuCl}_2$ ,  $\text{PdCl}_2$ ,  $\text{PtCl}_2$ ,  $\text{HgCl}_2$ ,  $\text{CdCl}_2$ ,  $\text{CuBr}_2$ ,  $\text{PdBr}_2$ ,  
 $\text{PtBr}_2$ ,  $\text{HgBr}_2$ ,  $\text{CdBr}_2$ )

Adams D.M., Goldstein M., Mooney E.F.

Trans. Faraday Soc., 1963, 59, N 10, 2228-32

Far-infra-red spectra of some anhydrous metal  
halides in the solid state.

PJF, 1964, 6D278

J.

ЕСТЬ ОРИГИНАЛ

rig.

PtCl<sub>2</sub>

Paptheodorou & N.

1979

J.  
cnekrp  
K.P.

Rept. 1978, CONF-780941-8

83 pp.

From Energy Res. Abstr.  
1979, 4(3), Abstr. 24088

(cell. PdCl<sub>2</sub>;  $\frac{1}{2}$ )

PtCl<sub>2</sub>

OM 30120

1988

Krebs B., Brendel C.,  
et al.,

(cnpyxrypa) Z. Anorg. und Allg.  
Chem., 1988, 561,  
Nr, 119- ● 131.